Acta Crystallographica Section E

## Structure Reports

Online
ISSN 1600-5368

R. Padmavathy, ${ }^{\text {a }}$ K. Saminathan, ${ }^{\text {a }}$<br>D. Natarajan, ${ }^{\text {b }}$ N. Bhavani ${ }^{\text {b }}$ and<br>K. Sivakumar ${ }^{\text {a* }}$<br>${ }^{\text {a }}$ Department of Physics, Anna University, Chennai 600025 , India, and ${ }^{\text {b }}$ Department of Chemistry, Annamalai University, Annamalai Nagar 608 002, India

Correspondence e-mail: surya@annauniv.edu

## Key indicators

Single-crystal X-ray study
$T=293 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.002 \AA$
Disorder in main residue
$R$ factor $=0.049$
$w R$ factor $=0.154$
Data-to-parameter ratio $=18.4$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0] Printed in Great Britain - all rights reserved

## 1-Acetyl-2-phenylperhydroquinolin-4-yl acetate

The piperidine and cyclohexane rings in the title compound, $\mathrm{C}_{19} \mathrm{H}_{25} \mathrm{NO}_{3}$, are trans-fused and adopt boat and chair conformations, respectively. The phenyl and ester substituents are oriented axially with respect to the decahydroquinoline moiety. The N atom of the piperidine ring exists in a planar configuration. The keto O atom of the ester group exhibits positional disorder, with an 80:20 population ratio. The crystal structure is stabilized by $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds.

## Comment

Decahydroquinolines are bicyclic systems consisting of a piperidine ring fused to a cyclohexane ring. The stereochemistry of decahydroquinolines is of interest as they occur in many alkaloids, such as lupinine, reserpine and yohimbine (Nasipuri, 1992). In decahydroquinolines, the cyclohexane ring has a preferred chair conformation, whereas the piperidine ring adopts a range of conformations, such as chair, twist or boat, depending upon the size and position of the substituents attached to the ring. The substitution of a methyl or nitroso group at the N -atom position has a large influence on the conformation of the ring (Vierhapper, 1980; Baliah \& Natarajan, 1989). Literature references show that decahydroquinoline derivatives exist with both cis- and trans-fused ring systems (Le Coz et al., 1990; Nierlich et al., 1991; Nogue et al., 1996; Ganesh et al., 1996; Thiruvalluvar et al., 1997; Paugam et al., 2000). Our study aims to investigate the nature of the ring fusion, the orientation of the substituents and their influence on the conformation of the piperidine ring. We present here the crystal structure of the title compound, (I) (Fig. 1).

(I)

One of the O atoms present in the ester group, O43, shows positional disorder, with an 80:20 population ratio. All the bond lengths and angles in this compound have values in agreement with other decahydroquinoline derivative structures.

Received 14 March 2005
Accepted 18 March 2005
Online 25 March 2005


Figure 1
A plot of the molecule of (I), with the atom-numbering scheme. Dislacement ellipsoids are drawn at the $50 \%$ probability level. Both disorder components of O43 are shown.


A perspective view of the molecule of (I), showing the chair and boat conformations of the fused rings.


Figure 3
The packing of the molecules of (I), viewed down the $b$ axis. Dashed lines indicate $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds.

The piperidine and cyclohexane rings of the decahydroquinoline moiety of (I) adopt a trans-fused geometry, with an $\mathrm{H} 9-\mathrm{C} 9-\mathrm{C} 10-\mathrm{H} 10$ torsion angle of 173 (1) ${ }^{\circ}$. The substituents at the N atom of the piperidine ring can yield either planar or pyramidal configurations (Ganesh et al., 1996; Thiruvalluvar et al., 1997; Paugam et al., 2000). The N atom of (I) has a planar configuration; relevant bond lengths and angles are given in Table 1.

Both the phenyl and the ester substituents at C 2 and C 4 are axial, with the relevant torsion angles listed in Table 1. While the cyclohexane ring exists in a chair conformation, the piperidine ring adopts a boat conformation. Fig. 2 gives a clear picture of the conformations of both rings.

The crystal structure of (I) is stabilized by $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds involving both keto O atoms, O 12 and O 43 (Table 2), and by van der Waals interactions

## Experimental

The title compound was prepared according to the procedure of Baliah et al. (1978). The white precipitate which was obtained was filtered and recrystallized from ethanol.

## Crystal data

$\mathrm{C}_{19} \mathrm{H}_{25} \mathrm{NO}_{3}$
$M_{r}=315.40$
Triclinic, $P \overline{1}$
$a=8.856(1) \AA$
$b=9.050(2) \AA$
$c=12.343(1) \AA$
$\alpha=85.35(1)^{\circ}$
$\beta=87.82(1)^{\circ}$
$\gamma=64.82(1)^{\circ}$
$V=892.3(2) \AA^{\circ}$

$$
\begin{aligned}
& Z=2 \\
& D_{x}=1.174 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation } \\
& \text { Cell parameters from } 25 \\
& \quad \text { reflections } \\
& \theta=10-20^{\circ} \\
& \mu=0.08 \mathrm{~mm}^{-1} \\
& T=293(2) \mathrm{K} \\
& \text { Prism, colourless } \\
& 0.42 \times 0.38 \times 0.30 \mathrm{~mm}
\end{aligned}
$$

## Data collection

Enraf-Nonius CAD-4 diffractometer
$\omega / 2 \theta$ scans
Absorption correction: none
4839 measured reflections
4056 independent reflections
2925 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.013$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.049$
$w R\left(F^{2}\right)=0.154$
$S=1.11$
4056 reflections
221 parameters
H atoms treated by a mixture of independent and constrained refinement
$\theta_{\text {max }}=27.5^{\circ}$
$h=-1 \rightarrow 11$
$k=-11 \rightarrow 11$
$l=-16 \rightarrow 16$
2 standard reflections every 100 reflections intensity decay: $1 \%$

$$
\begin{aligned}
& w= 1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0836 P)^{2}\right. \\
&+0.0334 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
&(\Delta / \sigma)_{\max }<0.001 \\
& \Delta \rho_{\max }=0.19 \mathrm{e}^{-3} \\
& \Delta \rho_{\min }=-0.19 \mathrm{e}^{-3}
\end{aligned}
$$

Table 1
Selected geometric parameters ( $\left(\AA,^{\circ}\right)$.

| N1-C11 | $1.3565(17)$ | $\mathrm{N} 1-\mathrm{C} 9$ | $1.4827(18)$ |
| :--- | ---: | :--- | ---: |
| $\mathrm{N} 1-\mathrm{C} 2$ | $1.4707(16)$ |  |  |
| C11-N1-C2 | $122.98(12)$ | $\mathrm{C} 2-\mathrm{N} 1-\mathrm{C} 9$ | $117.65(10)$ |
| C11-N1-C9 | $116.86(11)$ |  |  |
| C9-N1-C2-C3 | $50.78(14)$ | $\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 10$ | $11.68(15)$ |
| $\mathrm{C} 9-\mathrm{N} 1-\mathrm{C} 2-\mathrm{C} 21$ | $-79.08(14)$ | $\mathrm{C} 2-\mathrm{N} 1-\mathrm{C} 9-\mathrm{C} 10$ | $1.57(15)$ |
| $\mathrm{N} 1-\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4$ | $-57.53(13)$ | $\mathrm{O} 41-\mathrm{C} 4-\mathrm{C} 10-\mathrm{C} 9$ | $-76.41(14)$ |
| C21-C2-C3-C4 | $70.23(15)$ | $\mathrm{C} 3-\mathrm{C} 4-\mathrm{C} 10-\mathrm{C} 9$ | $42.14(15)$ |
| C2-C3-C4-O41 | $131.39(12)$ | $\mathrm{C} 8-\mathrm{C} 9-\mathrm{C} 10-\mathrm{C} 4$ | $-173.21(12)$ |

Table 2
Hydrogen-bond geometry ( $\AA,{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{C} 3-\mathrm{H} 3 \mathrm{~B} \cdots \mathrm{O} 43^{\text {i }}$ | 0.97 | 2.59 | 3.521 (3) | 161 |
| $\mathrm{C} 3-\mathrm{H} 3 \mathrm{~B} \cdots \mathrm{O} 43^{\text {i }}$ | 0.97 | 2.56 | 3.403 (11) | 146 |
| $\mathrm{C} 25-\mathrm{H} 25 \cdots \mathrm{O} 12^{\text {ii }}$ | 0.93 | 2.50 | 3.322 (2) | 148 |
| $\mathrm{C} 3-\mathrm{H} 3 A \cdots \mathrm{O} 12^{\text {iii }}$ | 0.97 | 2.63 | 3.5340 (18) | 155 |
| $\mathrm{C} 13-\mathrm{H} 13 \mathrm{C} \cdots \mathrm{O} 12{ }^{\text {iii }}$ | 0.96 | 2.78 | 3.475 (2) | 130 |

The disorder in atom O43 was observed during refinement as a very large displacement ellipsoid. The position was split into two, and their occupancies were initially refined and later fixed at 80:20. The disordered atoms were given the same displacement paramters. The H atoms were located in difference Fourier maps. The positions of atoms H 9 and H 10 , which are attached to atoms C 9 and C 10 , were refined. All other H atoms were made to ride on their respective C atoms, with $\mathrm{C}-\mathrm{H}$ distances in the range $0.93-0.98 \AA$ and with $U_{\text {iso }}(\mathrm{H})$ $=1.2 U_{\mathrm{eq}}(\mathrm{C})$, or $1.5 U_{\mathrm{eq}}(\mathrm{C})$ for methyl atoms.

Data collection: CAD-4 EXPRESS (Enraf-Nonius, 1994); cell refinement: CAD-4 EXPRESS; data reduction: XCAD4 (Harms \& Wocadlo, 1995); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP3 (Farrugia, 1997) and PLATON (Spek, 2003); software used to prepare material for publication: SHELXL97.

The authors acknowledge the Regional Sophisticated Instrumentation Centre, Indian Institute of Technology, Chennai, for single-crystal data collection.

## References

Baliah, V. \& Natarajan, A. (1989). Indian J. Chem. A, 28, 609-611.
Baliah, V., Pandiarajan, K. \& Lakshmanan, M. R. (1978). Indian J. Chem. B, 16, 72-73.
Enraf-Nonius (1994). CAD-4 EXPRESS. Version 5.1/1.2. Enraf-Nonius, Delft, The Netherlands.
Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
Ganesh, V. K., Velmurugan, D., Thiruvalluvar, A., Parthasarathi, V., Ramasubbu, N. \& Natarajan, D. (1996). Acta Cryst. C52, 3177-3179.
Harms, K. \& Wocadlo, S. (1995). XCAD4. University of Marburg, Germany.
Le Coz, L., Veyrat-Martin, C., Wartski, L., Seyden-Penne, J., Bois, C. \& Philoche-Levisalles, M. (1990). J. Org. Chem. 55, 4870-4879.
Nasipuri, D. (1992). Stereochemistry of Organic Compounds, pp. 314-315. New Delhi: Wiley Eastern Limited.
Nierlich, M., Vigner, D., Lance, M., Nogue, D., Paugam, R. \& Wartski, L. (1991). Acta Cryst. C47, 1678-1680.

Nogue, D., Paugam, R., Wartski, L., Nierlich, M., Vigner, D. \& Lance, M. (1996). Acta Cryst. C52, 207-210.

Paugam, R., Valenciennes, E., Le Coz-Bardol, L., Garde, J., Wartski, L., Lance, M. \& Nierlich, M. (2000). Tetrahedron Asymm. 11, 2509-2523.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Gottingen, Germany.
Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.
Thiruvalluvar, A., Parthasarathi, V., Natarajan, D., Bhavani, N. \& Bhadbhade, M. (1997). Acta Cryst. C53, 1086-1088.

Vierhapper, F. W. (1980). J. Org. Chem. 45, 3111-3118.


[^0]:    (C) 2005 International Union of Crystallography

